

98. A Total Synthesis of (\pm)-Aphidicolin: Regio- and Stereoselective Conversion of 3 α ,18-Di-*O*-benzyl-17-nor-14-aphidicolen-16-one into (\pm)-Aphidicolin¹

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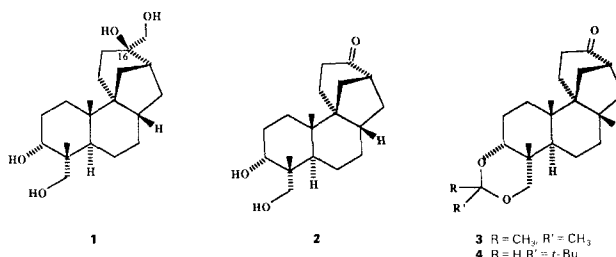
(5.IV.88)

A regio- and stereoselective conversion of the totally synthetic 3 α ,18-di-*O*-benzyl-17-nor-14-aphidicolen-16-one (**5**), into (\pm)-aphidicolin **1**, by hydroxylation of the 2-methylidenebicyclo[3.2.1]oct-3-ene derivative **6** is described. Compound **5** was a key intermediate in our previously described total synthesis of **2**, which represented a formal synthesis of **1**.

Introduction. – Although the diterpenic tetraol aphidicolin (**1**)² [1] attracted the attention of many synthetic chemists, only one stereoselective total synthesis of this interesting, biologically active [2] compound has been reported so far [3].

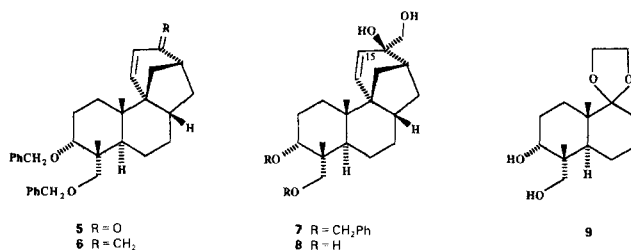
Since 17-noraphidicolan-16-one (**2**), a degradation product of **1**, has been reconverted *via* the acetone **3** into the natural product, albeit with poor stereoselectivity [1b], **2** [4b] or protected derivatives thereof, **3** [5–7] and **4** [8] [9], have been the targets of most synthetic plans. While such achievements constitute formal total syntheses of **1**, they do not address the problem of introducing the chiral centre at C(16), a hurdle by no means trivial [3] [6b] [8].

Other ingenious approaches to **1** [10] also do not avoid this difficulty. When the conversion of **2**, **3**, or **4** into **1** was accomplished, the lack of stereochemical control, necessitated a difficult separation of **1** from its C(16)-epimer [1b] [6b] [8].



¹) This work was presented at the IASOC II Meeting, Ischia, May 1986.

²) Systematic name: *tetradecahydro-3,9-dihydroxy-4,11b-dimethyl-8,11a-methano-11aH-cyclohepta[a]naphthalene-4,9-dimethanol.*



The absence of stereoselectivity may be attributed in large part to the conformational mobility of the bicyclo[3.2.1]octan-2-one³⁾ system, which constitutes the C/D ring moiety of these intermediates.

During the total synthesis of **2**, which we described some years ago [4b], we observed that the C/D ring system of the diene **6**, readily prepared from the enone **5**, has a rigid norcamphor-like conformation which, restricting the approach of reagents from the *exo*-side, should allow its stereoselective conversion to **1**.

The documented *exo*-reactivity of geometrically related systems [14] supported this hypothesis. In this paper, we report the successful implementation of this strategy in the conversion of intermediate **5** to (±)-aphidicolin²⁾.

Results and Discussion. – Compound **5** was converted quantitatively into the diene **6**, by reaction with MeLi in THF at 0°, followed by dehydration with TsOH in refluxing benzene. Compound **6** was then hydroxylated with OsO₄ and the osmate cleaved *in situ* with NaHSO₃ to give, as the only detectable reaction product, regio- and stereoselectively the diol **7**⁴⁾.

The configuration at C(16) of **7**, predictable on the basis of the precedents mentioned above [14], followed from its conversion into (±)-**1**. Compound **7** was then debenzylated by catalytic hydrogenation in the presence of 10% Pd(C) yielding quantitatively the unsaturated tetraol **8**.

The C(14)=C(15) double bond, which also previously played a key role in our synthesis by steering the rearrangement [4b], was finally reduced under pressure in the presence of 5% Rh/Al₂O₃ [16] to give (±)-aphidicolin indistinguishable from an authentic sample on the basis of chromatographic as well as spectroscopic comparisons. The described reaction sequence from **5** to **1** can be carried out without any chromatographic separation.

Conclusions. – The synthesis of **5** reported previously [4b], in connection with the work described herein, constitutes an efficient total synthesis of racemic (±)-**1** in which complete regio- and stereocontrol are maintained throughout.

The availability of the known precursor diol **9** [5] [6] in optically active form [17] makes this an attractive route for enantioselective synthesis of aphidicolin.

³⁾ Notwithstanding many attempts [6b] [8] [11] [12], the direct stereoselective nucleophilic addition to a bicyclo[3.2.1]octan-2-one system has been accomplished, to our knowledge, only in the synthesis of cedrol, where one side of the CO group was hindered [13]. A related strategy, exploiting steric encumbrance, was used to prepare (±)-**1** stereoselectively from a 2-methylidenebicyclo[3.2.1]octane [3].

⁴⁾ An analogous reaction was performed in the course of the total synthesis of maritimidol [15].

Experimental Part

General. See [2] [3]. UV: *Varian 634*. $^1\text{H-NMR}$: *Varian XL-300* spectrometer. HR-MS: *Kratos MS 80 RF*. R. P. > 7500, T = 190°, E. E. 70 eV. ψ = pseudo.

3 α ,18-Di-O-benzyl-16-methylidene-14-aphidicolene (6). To a stirred soln. of the enone **5** (25 mg, 0.05 mmol) in anh. Et₂O (5 ml), cooled in an ice-bath, was added dropwise a 1.6 M soln. (0.1 ml, 3 times excess) of CH₃Li in Et₂O under N₂. When TLC monitoring (Et₂O/petroleum ether (40–70°) 1:1; R_f (product) < R_f (**5**)) indicated completion of the reaction, H₂O (1 ml) was added dropwise and the whole thoroughly extracted with Et₂O. Combined org. extracts were then washed with H₂O until neutral, with brine, dried (Na₂SO₄), and evaporated. The residue, dissolved in dry benzene (15 ml), was then refluxed for 1 h in the presence of a catalytic amount of TsOH. When TLC monitoring indicated the disappearance of the starting material, the mixture was cooled to r.t., diluted with Et₂O, poured into a separatory funnel, and washed with an aq. NaHCO₃ soln., H₂O, brine, dried (Na₂SO₄), and evaporated affording, in an overall yield of 95% from **3**, the diene **6** (TLC: Et₂O/petroleum ether (40–70°) 5:95; R_f (**6**) > R_f (start. mat.)). Compound **6** was sufficiently pure to be used in the next reaction without purification. For anal. purposes, a sample was crystallized from Et₂O/CH₃OH. M.p. 89–90°. UV (EtOH): 237 (1.76 × 10⁴). IR (CCl₄): 2930, 2860, 1635, 1452, 1095. $^1\text{H-NMR}$ (CDCl₃): 1.03 (s, 3H); 1.05 (s, 3H); 2.75 (ψ t, 1H); 3.28 (d, J = 7.9, 1H); 3.55 (ψ t, 1H); 3.56 (d, J = 7.8, 1H); 4.38 (B of AB, J_{AB} = 11.7, 1H); 4.44 (B of AB, J_{AB} = 12.3, 1H); 4.45 (d, J = 1.90, 1H); 4.47 (A of AB, J_{AB} = 12.2, 1H); 4.58 (d, J = 1.8, 1H); 4.60 (A of AB, J_{AB} = 11.9, 1H); 5.99 (dd, J = 9.9, 1.5, 1H); 6.20 (d, J = 9.9, 1H); 7.20–7.50 (m, 10H). MS: 91 (100), 188 (7), 226 (8), 268 (26), 390 (4), 482 (0.15). MS (HR): 482.3193 (C₃₄H₄₂O₂, M⁺; calc. 482.3184).

3 α ,18-Di-O-benzyl-14-aphidicolene (7). To a soln. (1.5 ml) of **6** (14 mg, 0.03 mmol) in pyridine/Et₂O (1.5 ml/50 ml) cooled to –20°, was added a soln. (1.6 ml) of OsO₄ in Et₂O (100 mg/20 ml) precooled to the same temp. The whole was kept at –20° in the dark (freezer) and stirred from time to time for 2 days, till TLC monitoring (petroleum ether (40–70°)/Et₂O 95:5) indicated the disappearance of the starting material. The org. solvent was then evaporated, the residue taken up with 95% EtOH (3 ml), and a 0.3M aq. soln. (2 ml) of Na₂S₂O₅ was added. The brown mixture was stirred at r.t. for 2 h. The whole was then thoroughly extracted with CHCl₃. Colorless combined org. extracts were then washed with H₂O, brine, dried (Na₂SO₄), and evaporated to give crude **7**. For anal. purposes, a sample was crystallized from Et₂O/hexane. M.p. 99–100°. The yield was 83%. IR (CCl₄): 3630, 3590, 3410, 2950, 2870, 1620, 1455, 1100. $^1\text{H-NMR}$ (CDCl₃): 1.04 (s, 3H); 1.05 (s, 3H); 3.26 (d, J = 7.8, 1H); 3.54 (d, J = 7.3, 1H); 3.56 (B of AB, J_{AB} = 11.0, 1H); 3.63 (d, J = 11.2, 1H); 4.38 (d, J = 11.5, 1H); 4.44 (B of AB, J_{AB} = 12.2, 1H); 4.47 (A of AB, J_{AB} = 12.2, 1H); 4.58 (A of AB, J_{AB} = 11.5, 1H); 5.50 (dd, J = 10.3, 1.5, 1H); 6.25 (dd, J = 8.8, 1.5, 1H); 7.1–7.5 (m, 10H). MS: 91 (100), 255 (9), 271 (4), 284 (5), 498 (1). HR-MS: 498.3127 (C₃₄H₄₂O₃, M⁺ – 18; calc. 498.3127).

14-Aphidicolene (8). A soln. of **7** (5 mg, 0.015 mmol) in 95% EtOH (3 ml) was hydrogenated, at atmospheric pressure and r.t., in the presence of a catalytic amount of 10% Pd/C (*Fluka AG*). When TLC monitoring (Et₂O/petroleum ether (40–70°)/CH₃OH 7.5:2:0.6) indicated the disappearance of the starting material, the catalyst was removed by filtration through a *Celite* pad, which was thoroughly washed with 95% EtOH. Evaporation of the solvent afforded quantitatively **8**, which was crystallized from CHCl₃/Et₂O. M.p. of 148–150°. IR (CHCl₃): 3420, 2940, 1605, 1090. $^1\text{H-NMR}$ (CDCl₃): 0.73 (s, 3H); 1.03 (s, 3H); 3.15 (ψ s, 1H); 3.30–3.80 (m, 6H); 5.54 (dd, J = 10.0, 1, 1H); 6.31 (dd, J = 10.1, 1, 1H); MS: 79 (22), 91 (29), 105 (20), 107 (20), 145 (11), 147 (10), 187 (19), 305 (100), 318 (0.5). HR-MS: 305.2117 (C₁₉H₂₉O₃, M⁺ – 31; calc. 305.2117).

(±)-*Aphidicolin (1).* A stirred soln. of **8** (5 mg, 0.014 mmol) in abs. EtOH (4 ml) was hydrogenated in the presence of a catalytic amount of 5% Rh/Al₂O₃ (*Engelhard Industries Inc.*) for 24 h at 1000 psi and at r.t. in a steel *Parr* apparatus. The catalyst was then removed by filtration through a *Celite* pad, which was thoroughly washed with 95% EtOH. Evaporation of the solvent and purification by SiO₂-CC afforded in 73% (±)-**1** which was indistinguishable from an authentic sample by TLC (Et₂O/petroleum ether (40–70°)/CH₃OH 8:2:0.6, 3 developments, R_f (**8**) > R_f (**1**) [6b]⁵). $^1\text{H-NMR}$ (CDCl₃): 0.71 (s, 3H); 0.99 (s, 3H); 3.38 (B of AB, J_{AB} = 11.4, 2H); 3.47 (A of AB, J_{AB} = 11.2, 2H); 3.69 (ψ s, 1H). MS: 180 (29), 181 (9), 217 (7), 231 (6), 259 (8), 271 (7), 275 (8), 289 (4), 290 (4), 307 (100), 308 (15), 320 (0.1). HR-MS: 307.2271 (C₁₉H₃₁O₃, M⁺ – 31; calc. 307.2273).

⁵ Under these chromatographic conditions, C(16)-epi-aphidicolin displays an R_f value < 1 [6b].

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